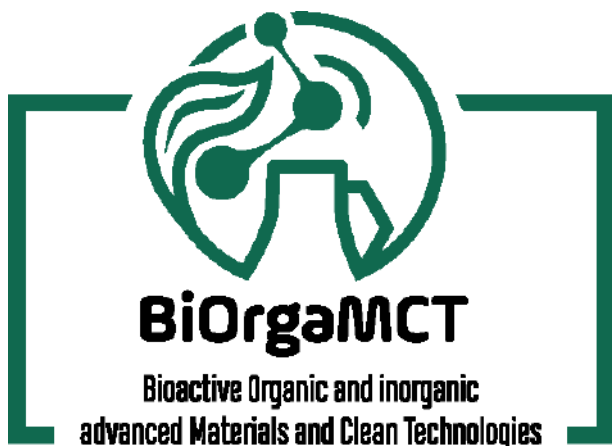




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MINISTRY
OF EDUCATION
AND SCIENCE

Organic Functional Materials
Part of the project №BG-RRP-2.004-0002, "BiOrgaMCT"

9 October 2025, Sofia

Laboratory of Organic Functional Materials

WP 1 Synthesis of Organic Materials

- *Design of new molecules*
- *Synthesis and purification*
- *Structure elucidation*

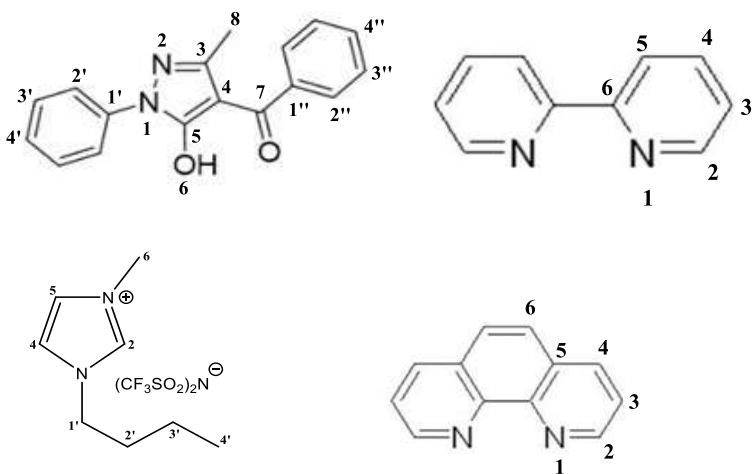
WP 2 Investigation of the Photophysical Properties

- *Steady-state and transient spectroscopy measurements*
- *Dynamic spectral measurements upon external stimuli*
- *Preparation of 1D and 2D solid-state materials*

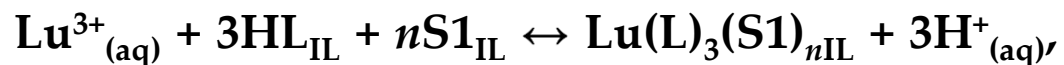
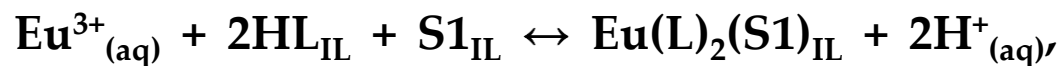
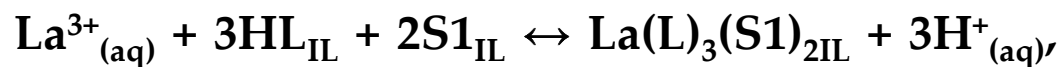
WP 3 Study and Evaluation their Working Functions as Molecular Devices

- *Solid-state performance of organic materials (thin films, bulk material, deposition of various substrates)*
- *Investigation microscopic characteristics by TEM, SEM, XRD*
- *Structure-properties relationship evaluation*

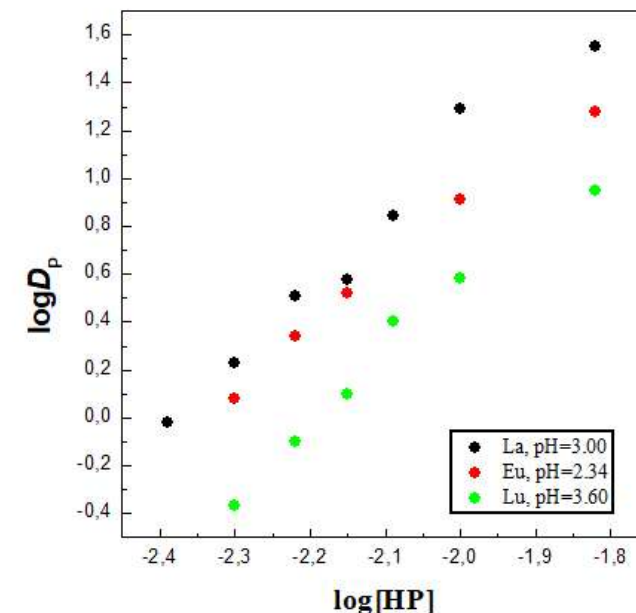
SOLVENT EXTRACTION OF METALS WITH FANCY LIGANDS



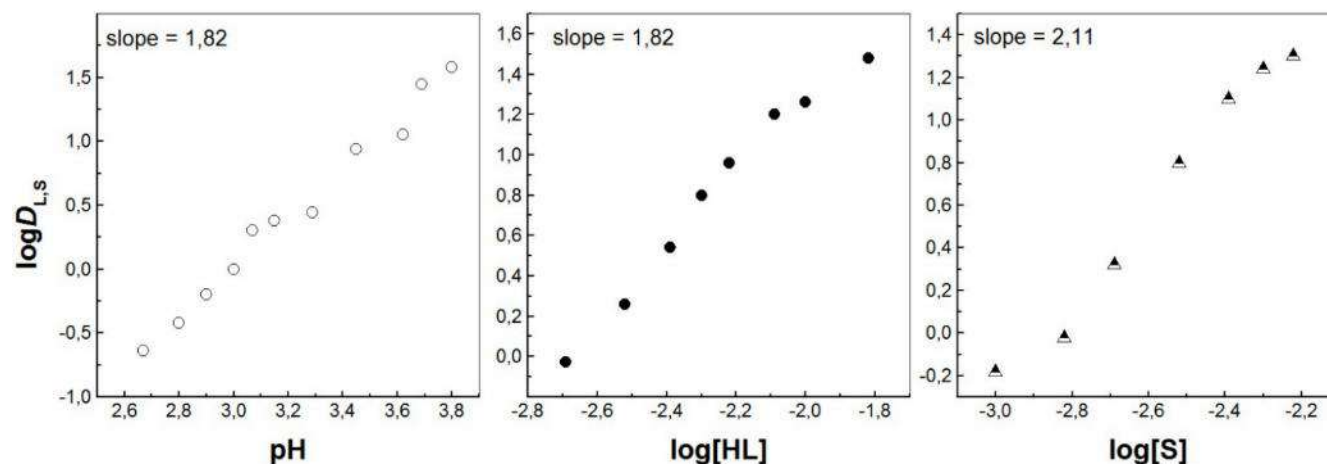
2 > 1 → synergism



Мултифункционално химично моделиране на екстракция с йонна течност на различни метали с различни леганди с цел подобряване на ефективността на процеса и неговата екологичност. Приложен е и нетрадиционен синергентен подход с изследване ефекта на комбинация от два лиганда.



Log D vs. log $[\text{HP}]_{\text{IL}}$ plots at constant pH for La^{3+} , Eu^{3+} and Lu^{3+} in $[\text{C}_1\text{C}_4\text{im}^+][\text{Tf}_2\text{N}^-]$.



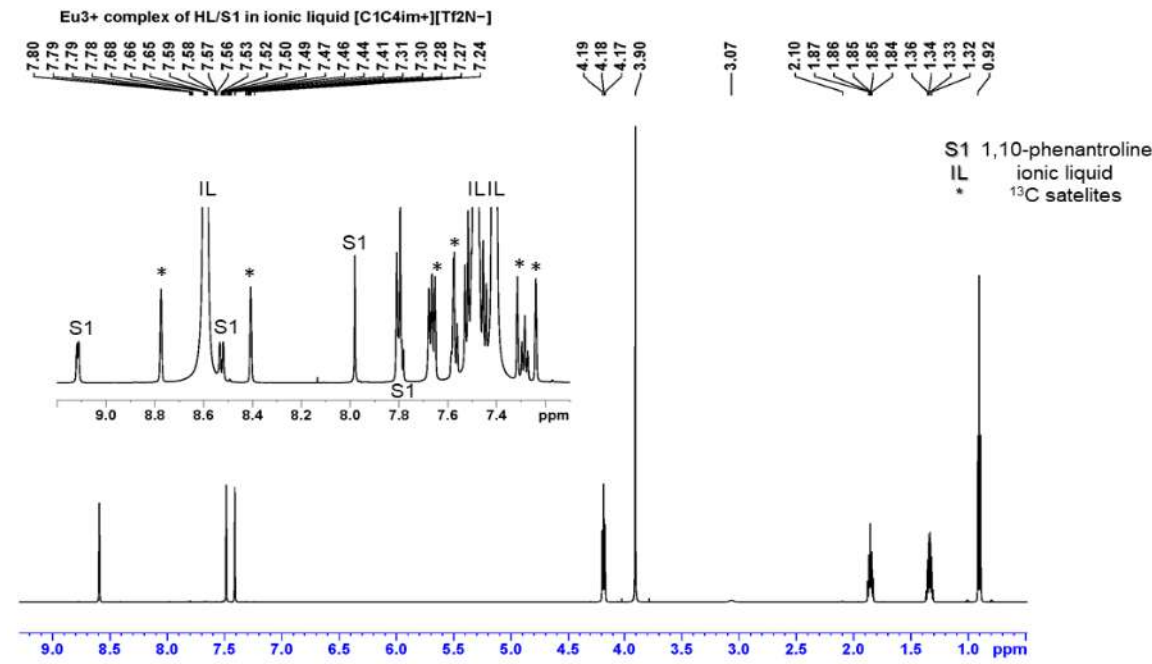
Log $D_{\text{L,S}}$ vs. log pH ($[\text{HL}] = 5 \times 10^{-3} \text{ mol/dm}^3$ and $[\text{S}] = 3 \times 10^{-3} \text{ mol/dm}^3$), log $D_{\text{L,S}}$ vs. log $[\text{HL}]_{\text{IL}}$ ($[\text{S}] = 3 \times 10^{-3} \text{ mol/dm}^3$ at pH=3) and log $D_{\text{L,S}}$ vs. log $[\text{S}]$ ($[\text{HL}] = 5 \times 10^{-3} \text{ mol/dm}^3$ at pH=2.78) plots for solvent extraction of La^{3+} with HL–bipy mixture in $[\text{C}_1\text{C}_4\text{im}^+][\text{Tf}_2\text{N}^-]$.

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Li(I)										IL HP S1 S2 HP-S1 HP-S2				
Na(I)	Mg(II)											Al(III)		
K(I)	Ca(II)	Sc(III)	Ti(IV)	V(V)	Cr(III)	Mn(II)	Fe(III)	Co(II)	Ni(II)	Cu(II)	Zn(II)	Ga(III)		
6.2%							15.5%			1.9%				
8.2%							99.3%			87.5%				
1.6%							16.9%	100%	89.3%	99.8%				
							16.6%	100%	99.7%	99.8%				
							98.4%	100%	57.4%	100%				
							98.2%	99.05%	99.7%	100%				
Rb(I)	Sr(II)	Y(III)	Zr(IV)	Nb(V)	Mo		Ru(III)	Rh(III)	Pd(II)	Ag(I)	Cd(II)	In(III)		
										99.7%				
										99.7%				
										99.7%				
										100%				
										100%				
Cs(I)	Ba(II)	Ln(III)	Hf(IV)	Ta(V)	W(VI)	Re(VII)	Os(IV)	Ir(III)	Pt(IV)	Au(III)	Hg(II)	Tl(I)	Pb(II)	Bi(III)
											17.2%		21.09%	99.8%
											11.2%		4.2%	99.8%
											86.6%		4.0%	99.8%
											28.3%		3.09%	99.8%
											70.5%		3.9%	100%
											35.9%		1.5%	100%
Ln(III)														
La(III)	Ce(III)	Pr(III)	Nd(III)	Sm(III)	Eu(III)	Gd(III)	Tb(III)	Dy(III)	Ho(III)	Er(III)	Tm(III)	Yb(III)	Lu(III)	
4.3%	8.3%				39.07%								42.6%	
1.05%	1.9%				1.5%								1.9%	
4.07%	7.04%				44.6%								49.7%	
6.0%	8.3%				44.3%								48.4%	



pH_{ex} = 1.65

Понастоящем ролята на лигандите е научно изследвана поотделно и в контекста на един метал или една група метали, но без постигане на глобална картина и универсалност. Предложен е подход за екстракция „моментна снимка“, който е в рамките на екологична и устойчива химия, а и също ефективен прием в контекста на зелена индустрия с нулев отпадък.



Solvent extraction performance of IL, HL (1×10^{-2} mol/dm³), S1 and S2 (8×10^{-3} mol/dm³) ligands diluted in [C₁C₄im⁺][Tf₂N⁻] and as well as the two mixtures (HL-S1 and HL-S2: 1×10^{-2} mol/dm³ – 5×10^{-3} mol/dm³) for 22 metal ions

J. Mol. Liq. 415 (2024) 126332. Q1
Molecules 30 (2025) 786. Q2

¹H NMR spectrum of Eu³⁺ complex of HL/S1 in [C₁C₄im⁺][Tf₂N⁻] obtained during solvent extraction process, **without any dilution of the organic phase with an organic (deuterated) diluent.**

Light-driven self-sterilizing cotton fabric and drug delivery

Photochemical & Photobiological Sciences
https://doi.org/10.1007/s43630-025-00710-1

ORIGINAL PAPERS

Light-driven self-sterilizing cotton fabric and drug delivery: improvement of the antimicrobial activity of 4-sulfo-1,8-naphthalimide via its dendrimer and metallic dendrimer formation

Desislava Staneva¹ · Awad I. Said^{2,3} · Petar Grozdanov⁴ · Ivanka Nikolova⁴ · Radostina Stoyanova⁵ · Albena Jordanova² · Ivo Grabchev²

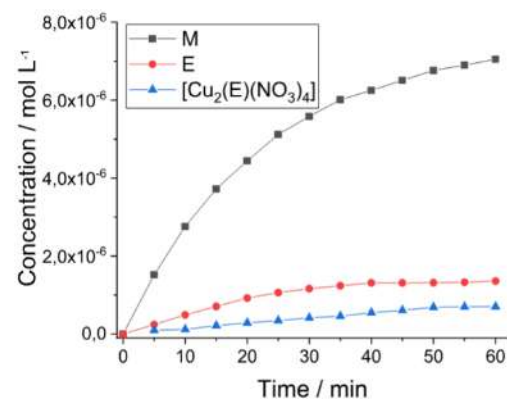
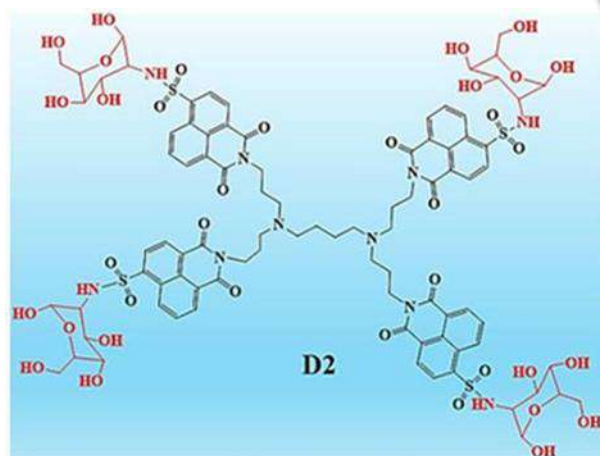
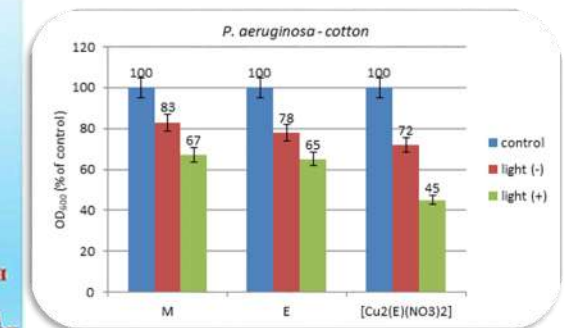
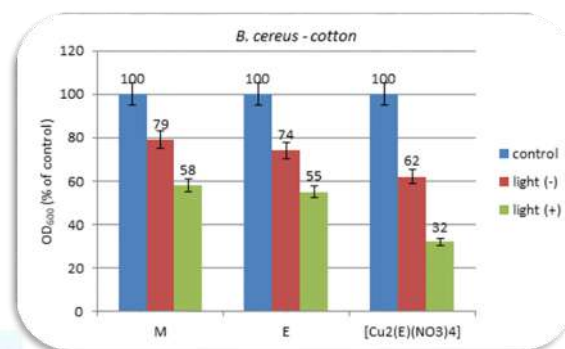
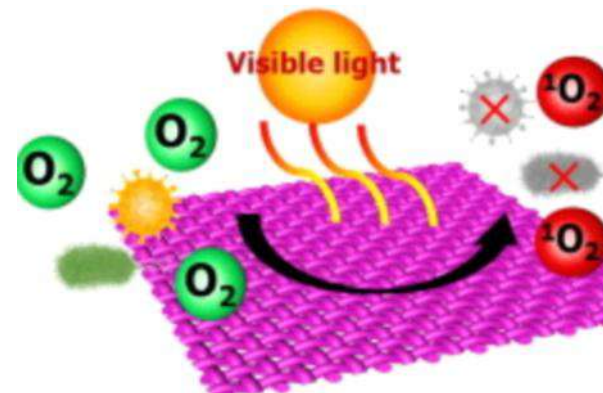
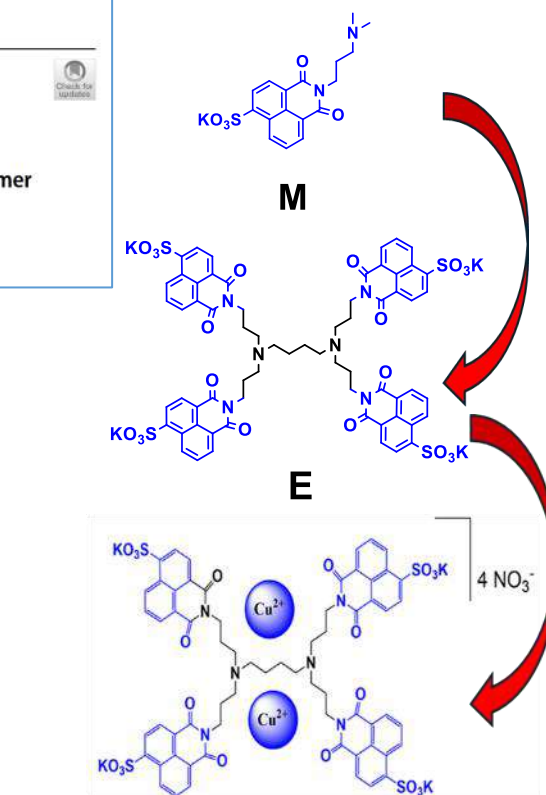
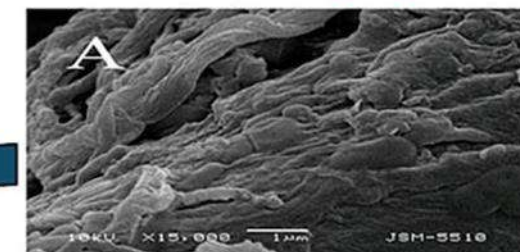


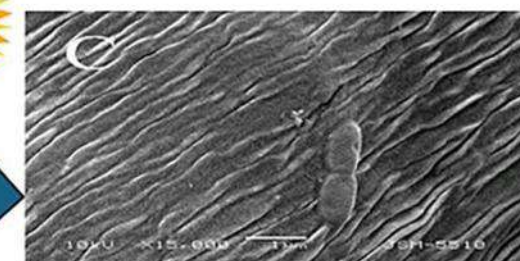
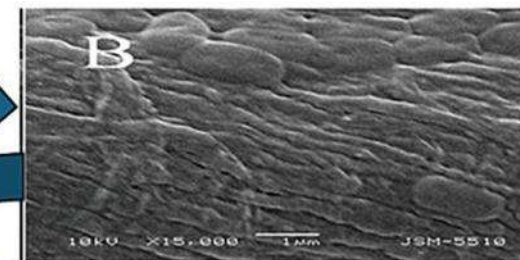
Figure. The release of monomer M, dendrimer E, and metallodendrimer [Cu₂(E)(NO₃)₄] from the cotton fabric in a phosphate buffer with pH=7.4 at 37 °C



SEM micrographs



biofilm of *P. aeruginosa* on cotton fabric



D2



Journal of Photochemistry & Photobiology, A: Chemistry 464 (2025) 116306

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A new photoactive water-soluble polypropylene imine dendrimer modified with 1,8-naphthalimide and N-glucosamine for light-driven self-sterilizing cotton fabrics

Awad I. Said^{a,b}, Desislava Staneva^{c,*}, Daniela Atanasova^c, Albena Jordanova^a, Ivo Grabchev^{a,c}

Textile with antimicrobial and cleaning properties



macromol



Article

Fluorescent Hyperbranched Polymers and Cotton Fabrics Treated with Them as Innovative Agents for Antimicrobial Photodynamic Therapy and Self-Disinfecting Textiles

Desislava Staneva ^{1,*}, Paula Bosch ², Petar Grozdanov ³, Ivanka Nikolova ³ and Ivo Grabchev ^{4,*}

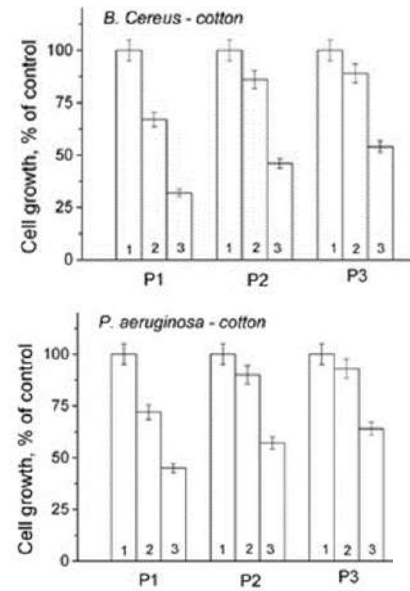
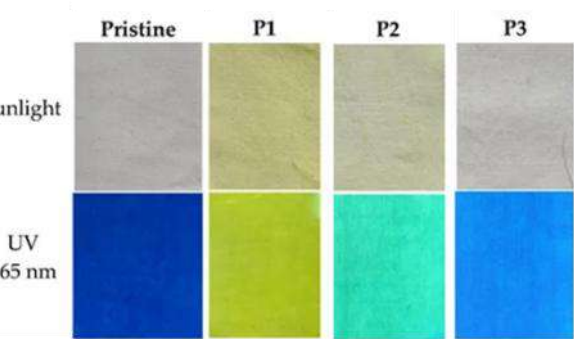
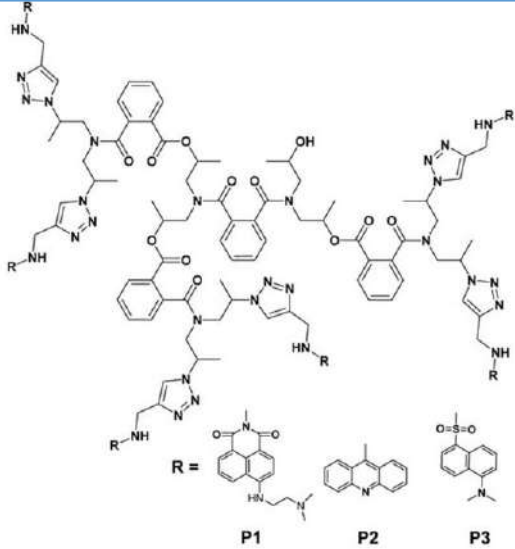


Figure. The growth of the model strains *P. aeruginosa* and *B. cereus* in the presence of cotton fabrics treated with polymers P1, P2, and P3, (1—control, 2—in the dark, and 3—after illumination).

Table. Virucidal effect of cotton fabric treated with the polymers against HRSV-2 and HAdV-5 (irradiated with light for 1 h and non-irradiated).

	HRSV-2				HAdV-5			
	Irradiated		Non-Irradiated		Irradiated		Non-Irradiated	
	Δlog 30 min	Δlog 60 min	Δlog 30 min	Δlog 60 min	Δlog 30 min	Δlog 60 min	Δlog 30 min	Δlog 60 min
P1	0	0.1	0	0	0	0.2	0	0
P2	0	0.1	0	0	0	0.1	0	0
P3	0	0.3	0	0	0	0.4	0	0



materials



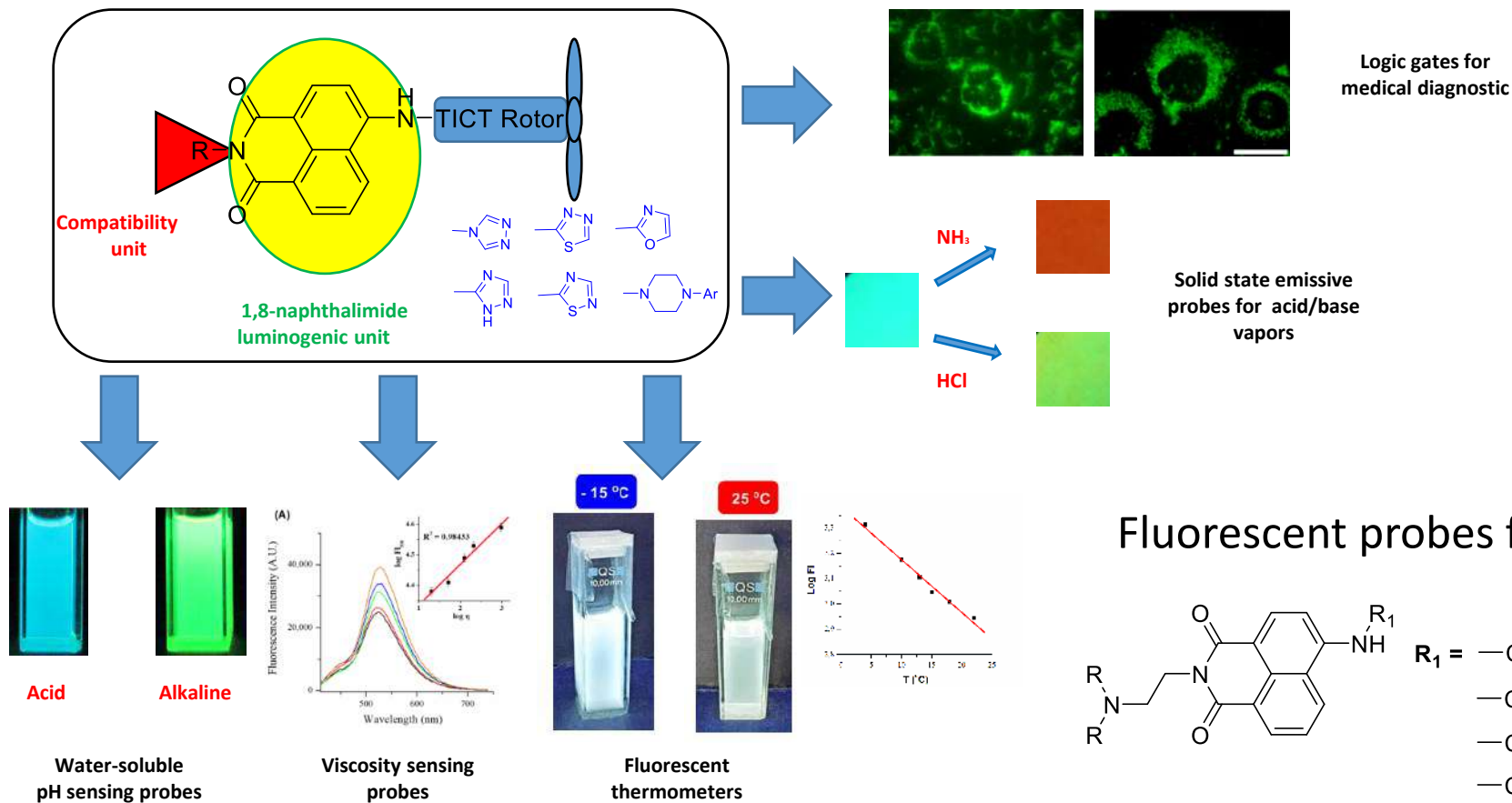
Article

Application of Textile Composite Materials as a Sorbent for Cleaning Up Oil Spills

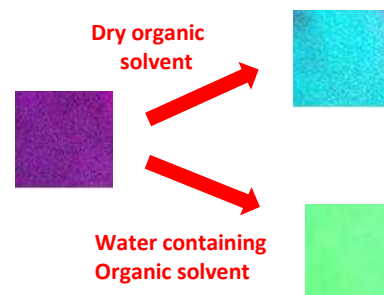
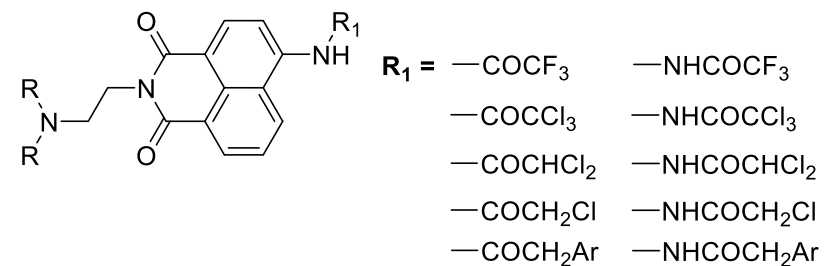
Daniela Angelova ^{*,}, Desislava Staneva ^{*,}, Daniela Atanasova and Vesislava Toteva ^{*,}



Fluorescent sensing 1,8-naphthalimide based TICT rotors



Fluorescent probes for humidity



Chiral and non-conjugated C2-spirooxindole substituted at the C3 position by a phenyl ring has been studied for its unusual photoluminescence behaviour.

Journal of Photochemistry & Photobiology, A: Chemistry 463 (2025) 116316



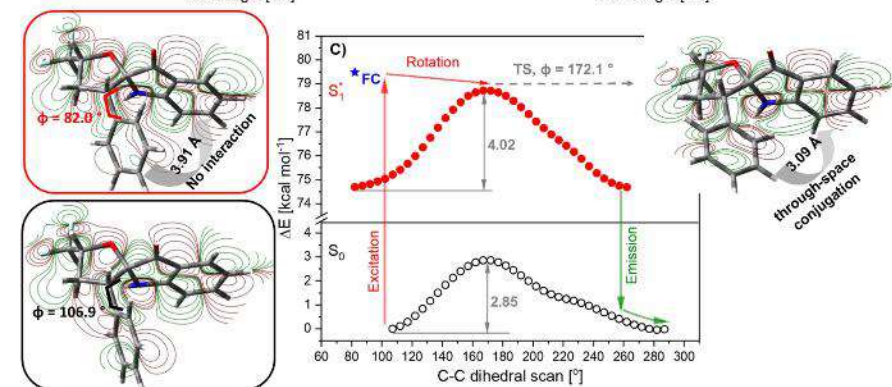
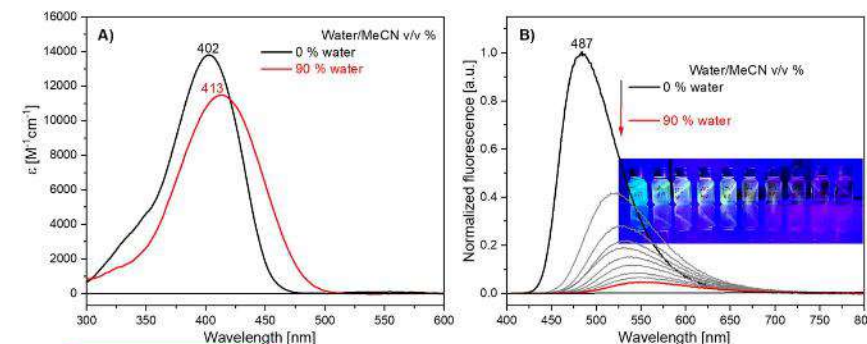
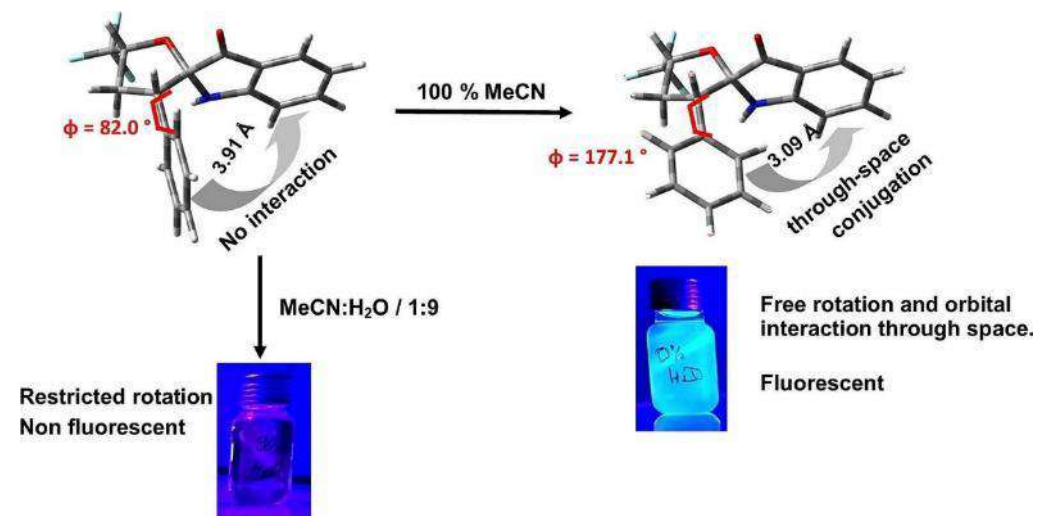
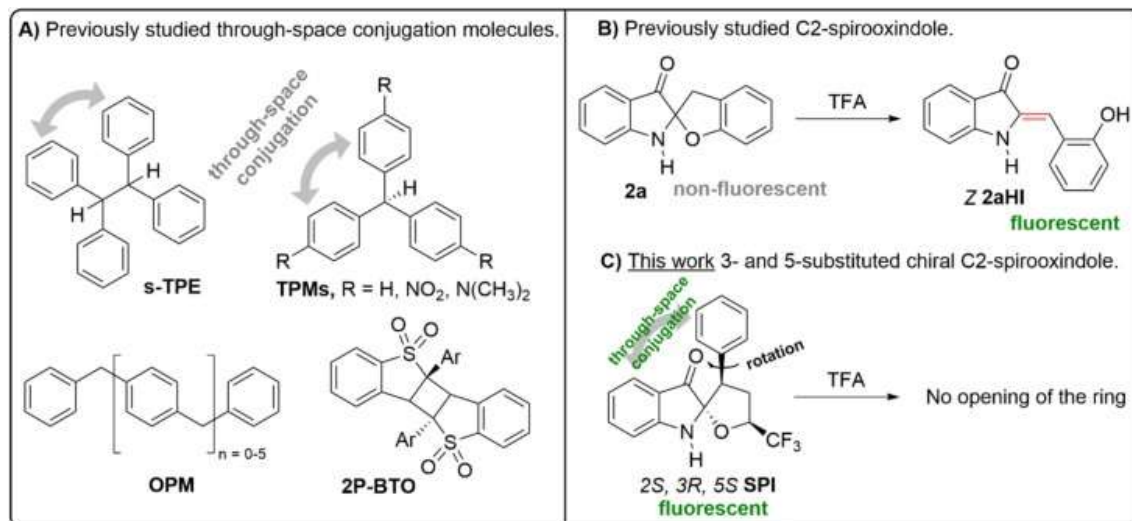
Intramolecular through-space conjugation of chiral C2-spirooxindole

Kosuke Nakashima^{a,*}, Aoi Imamura^a, Yasuyuki Matsushima^a, Shin-ichi Hirashima^a, Dancho Yordanov^{b,c}, Tsuyoshi Miura^a, Anton Georgiev^{b,*}

^a Department of Pharmaceutical Chemistry, Tokyo University of Pharmacy and Life Sciences, 1432-1 Horinouchi, Hachioji, Tokyo 192-0392, Japan

^b Department of Organic Chemistry, University of Chemical Technology and Metallurgy, 8 St. Kliment Ohridski blvd, 1756 Sofia, Bulgaria

^c Institute of Organic Chemistry with Centre of Phytochemistry, Bulgarian Academy of Sciences, Acad. G. Bonchev avenue, bldg. 9, Sofia 1113, Bulgaria



Visible Light Switching of *Ortho*-Functionalized Azo Phthalimides with Tunable Z-Isomer Stability

Dancho Yordanov, Kosuke Nakashima, Rastislav Smolka, Yasuyuki Matsushima, Shin-ichi Hirashima, Martin Vala, Tsuyoshi Miura, and Anton Georgiev*

Cite This: <https://doi.org/10.1021/acs.joc.5c01018>

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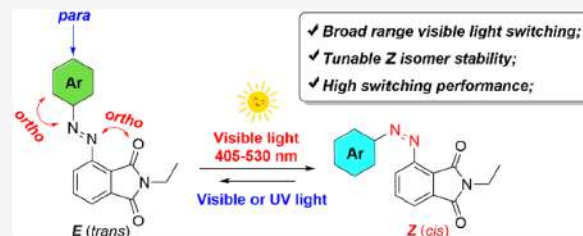
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Metrics & More

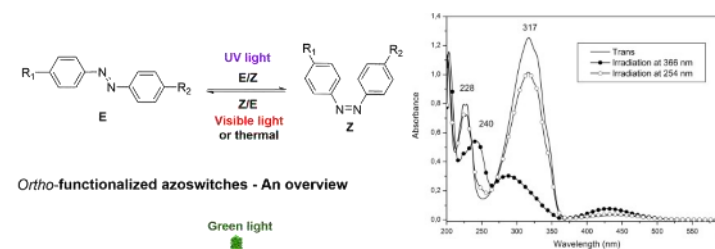
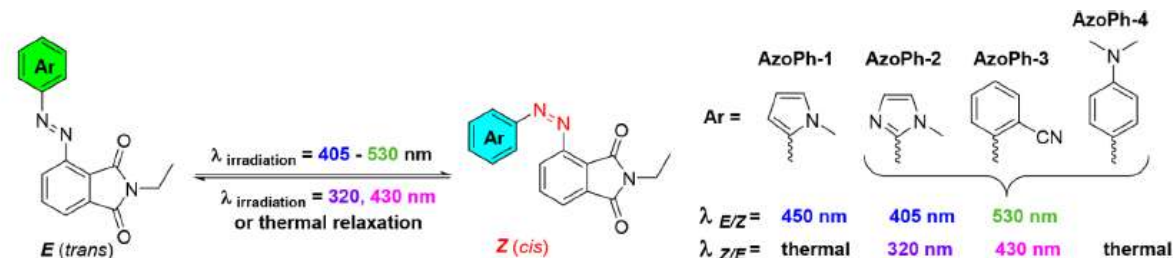
Article Recommendations

Supporting Information

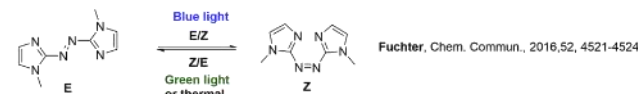
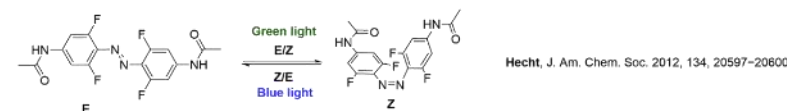
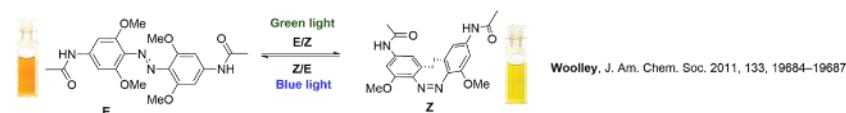
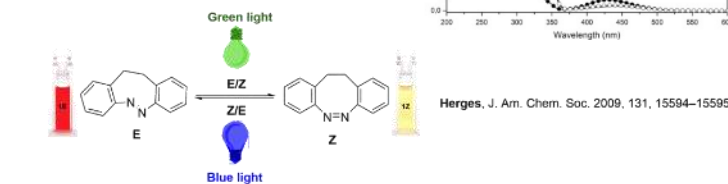
ABSTRACT: Herein, a novel class of azo photoswitches based on a phthalimide with an *ortho* azo bond to the imide ring is presented, exhibiting reversible isomerization under a broad range of visible light irradiation from 405 to 530 nm. Structural variations with heteroaryl or aryl segments attached to the 3-phthalylazo unit exhibit distinct spectral features, such as red-shifted absorption, well-separated absorption bands, and tunable stability of the metastable Z isomer, ranging from seconds to days. They differ drastically in the half-life of Z-isomer stability, ranging from several seconds (*N*-methylpyrrole) to days (*N*-methylimidazole). The *ortho*-cyanophenyl azo dye shows an $n-\pi^*$ band separation of 22 nm between the *E* and Z isomers, enabling on–off switching by green (530 nm) and blue (430 nm) light. X-ray structural analysis reveals that the *E* isomer adopts a planar geometry. The Z isomers exhibit a T-shaped configuration for dyes with heteroaryl segments, while aryl Z isomers adopt a twisted conformation. Ground-state density functional theory calculations suggest a different mechanism for Z/E thermal relaxation depending on the heteroaryl or aryl substituents, explaining the varying stability of the Z isomers. Van't Hoff analysis of the thermal Z/E isomerization provided enthalpic and Gibbs free energy changes, supporting the substituent-dependent trends in Z-isomer stability.



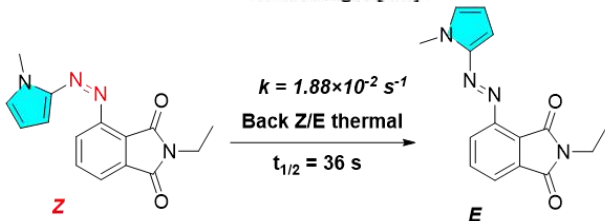
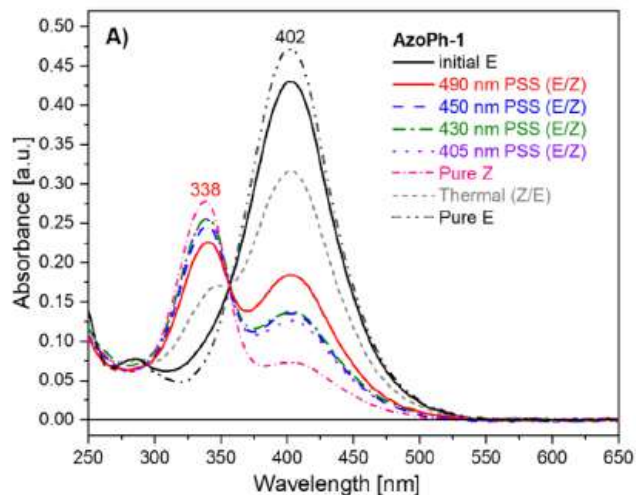
A novel class of azo photoswitches based on a phthalimide with an *ortho* substitution with improved characteristics has been synthesized and studied.



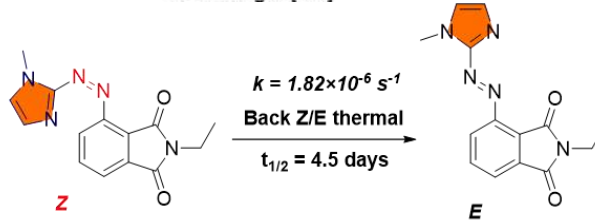
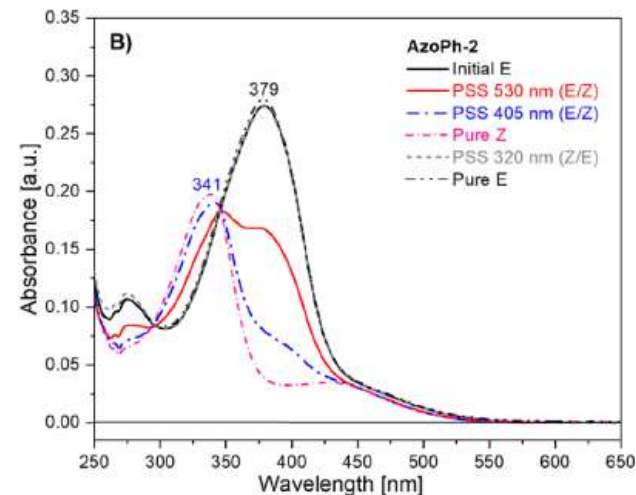
Poor $n-\pi^*$ band separation.
Impossible E/Z isomerization by visible light.



Molecular switches



AzoPh-1



AzoPh-2

The kinetics and thermodynamic comparative analysis reveals that AzoPh-1 and AzoPh-2 exhibit the highest ΔH° , indicating strong enthalpic stabilization of their Z isomers. However, only AzoPh-2 shows a positive ΔG° , confirming that the Z isomer is thermodynamically stable at ambient temperature and kinetically persistent, with the slowest thermal reversion ($t_{1/2} = 4.9$ days).

Plotting $\ln K$ vs $1/T$, yields:

$$\text{Slope: } -\frac{\Delta H^\circ}{R} \Rightarrow \Delta H^\circ = -\text{Slope} \times R$$

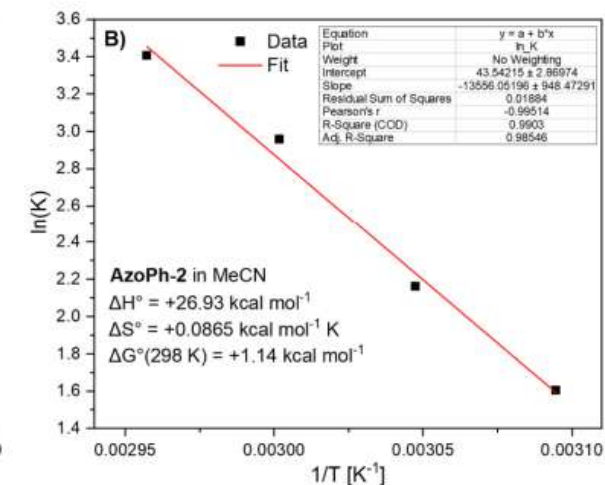
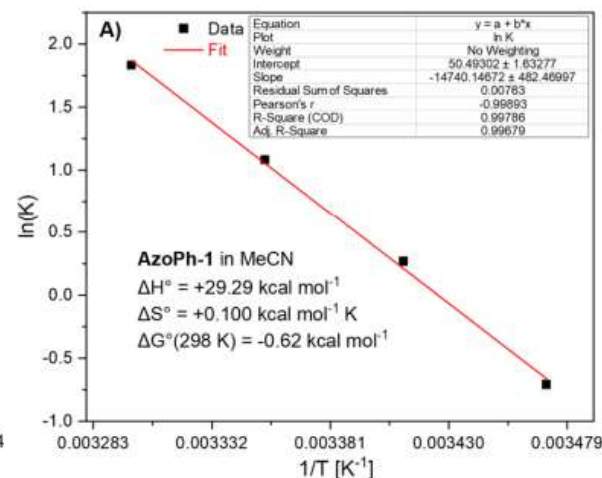
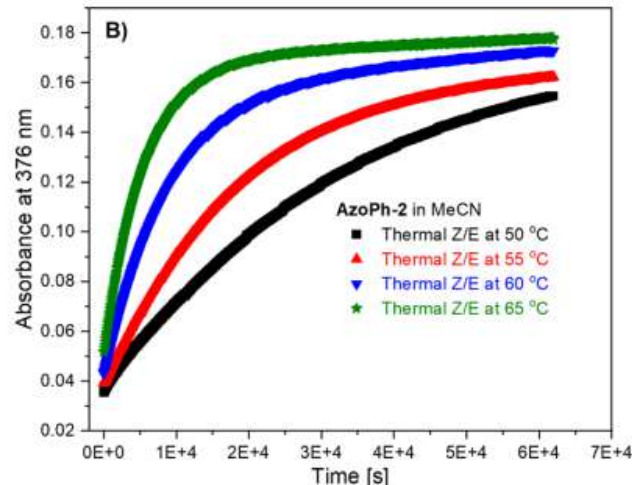
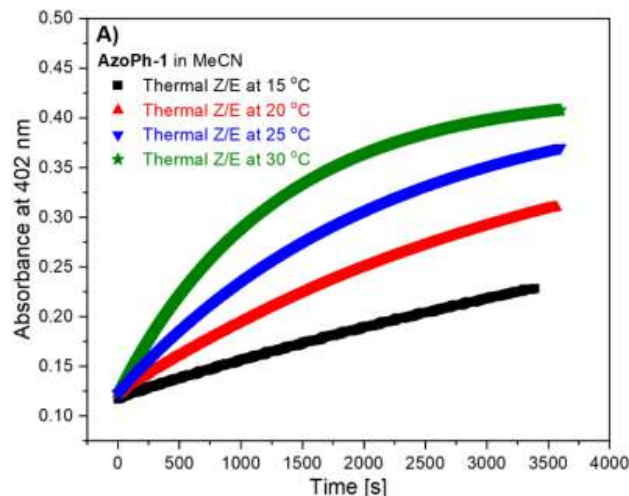
$$\text{Intercept: } +\frac{\Delta S^\circ}{R} \Rightarrow \Delta S^\circ = \text{Intercept} \times R$$

Then,

$$\Delta G^\circ = \Delta H^\circ - T\Delta S^\circ$$

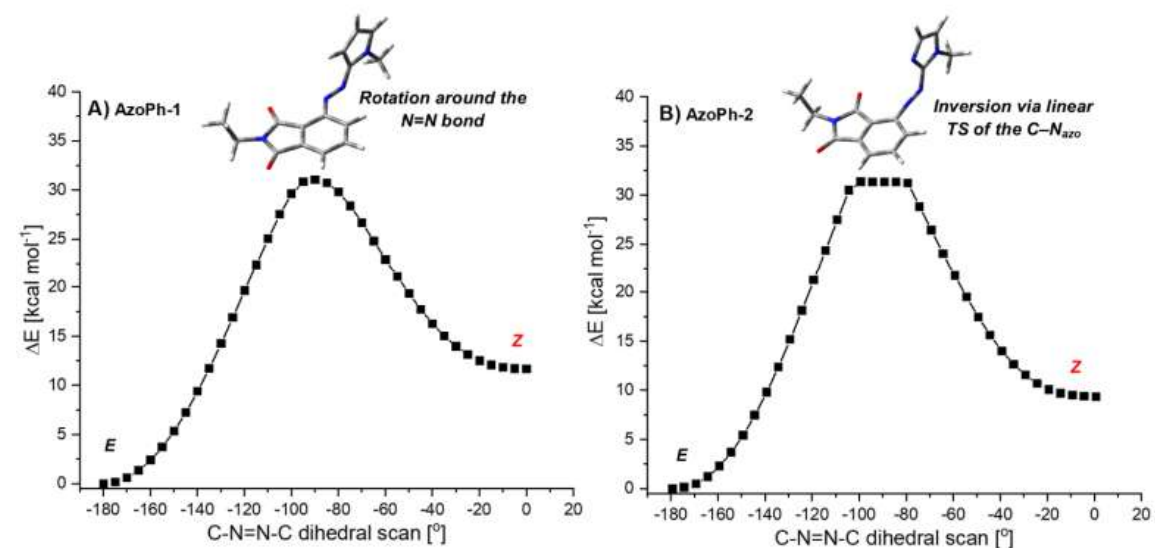
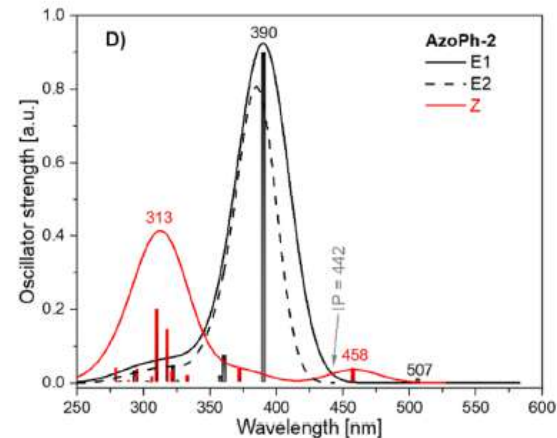
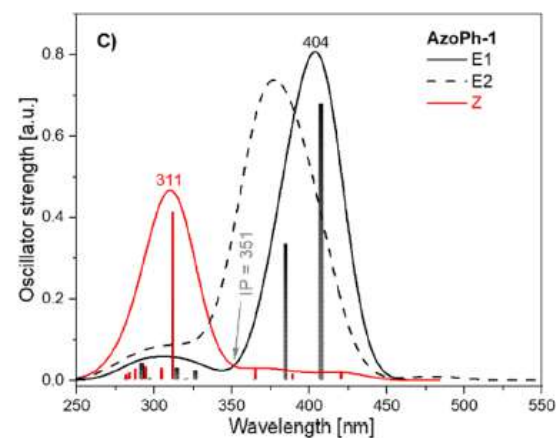
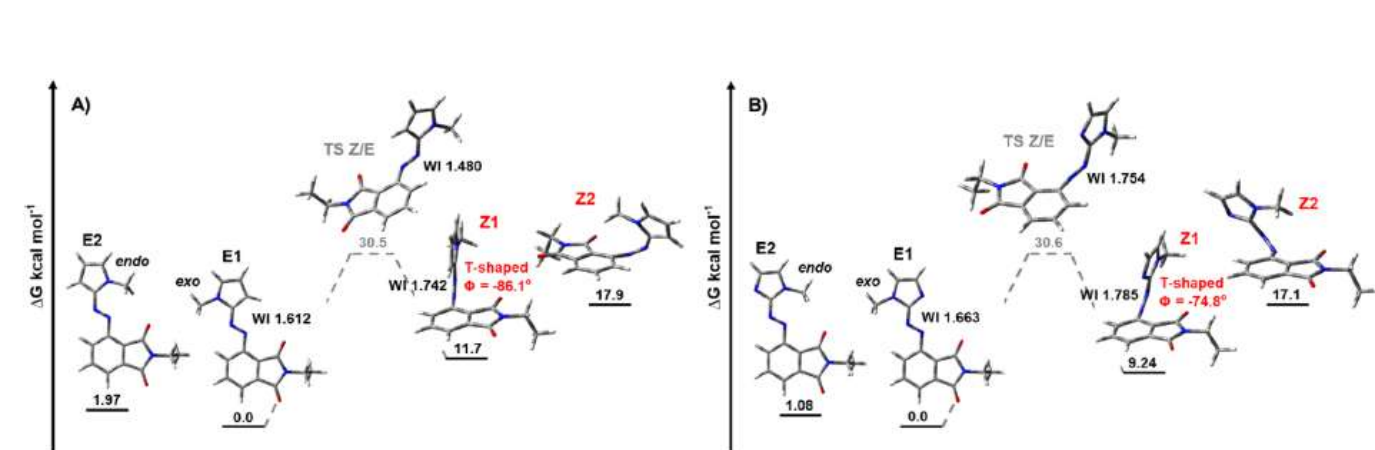
Van't Hoff analysis

$$\ln K = -\frac{\Delta H^\circ}{R} \times \frac{1}{T} + \frac{\Delta S^\circ}{R}$$



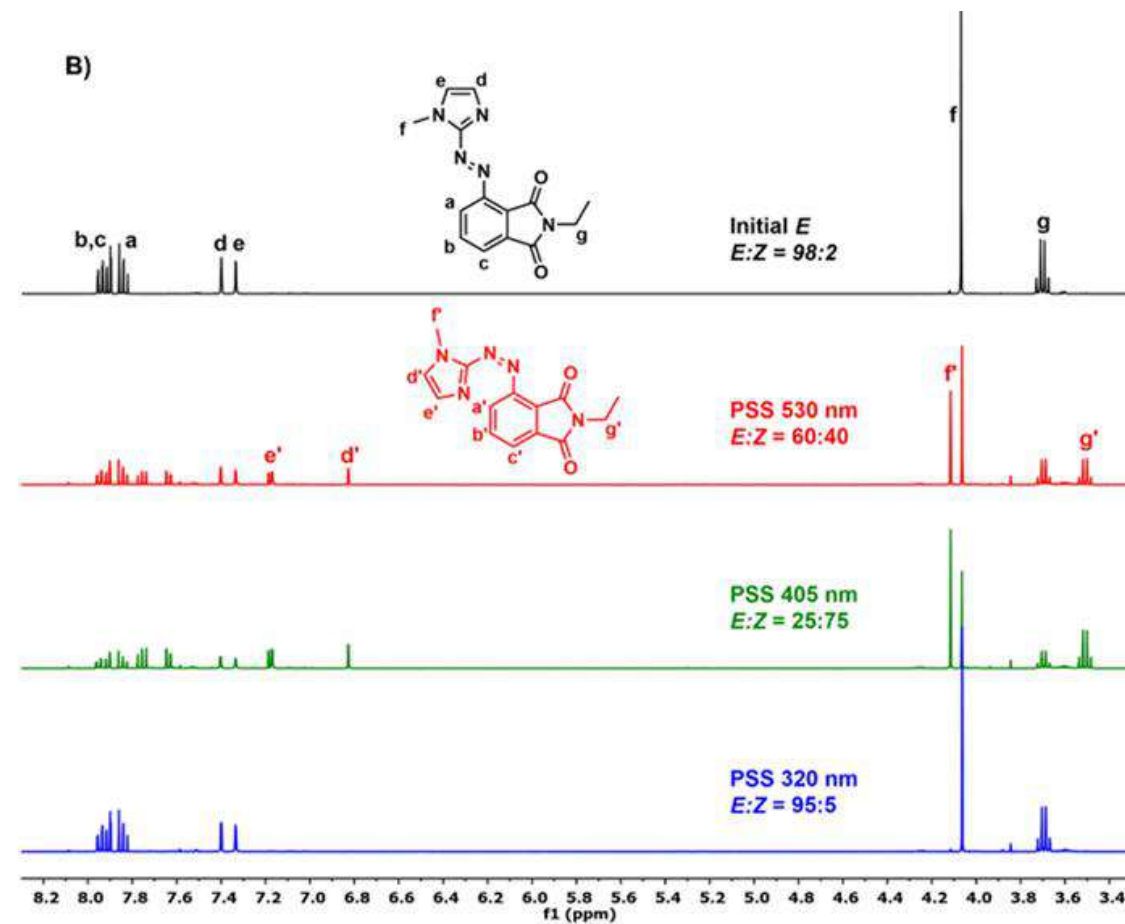
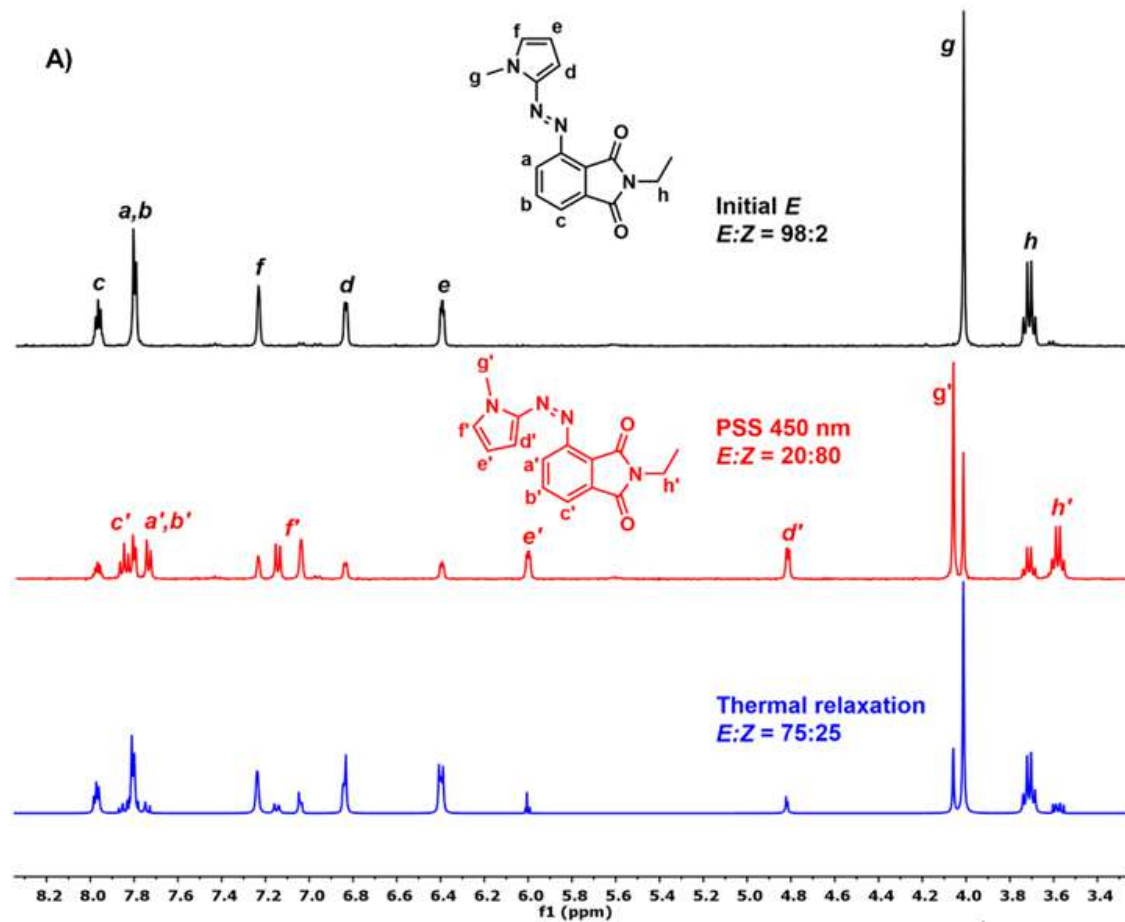
Molecular switches

Ground state DFT TS state geometry to establish the mechanism of thermal relaxation



Z/E potential energy profiles obtained by relaxed scans of the C-N=N-C dihedral angle

Molecular switches



¹H NMR (400 MHz, 25 °C, CD₃CN)

Anton Georgiev

1. Dancho Yordanova, Kosuke Nakashima, Rastislav Smolka, Yasuyuki Matsushima, Shin-ichi Hirashima, Martin Vala, Tsuyoshi Miura, Anton Georgiev, “Visible Light Switching of ortho-Functionalized Azo Phthalimides with Tunable Z-Isomer Stability” **The Journal of Organic Chemistry (American Chemical Society)** 2025, 90, 37, 12904–12915, <https://doi.org/10.1021/acs.joc.5c01018> Q1 (Web of Science)
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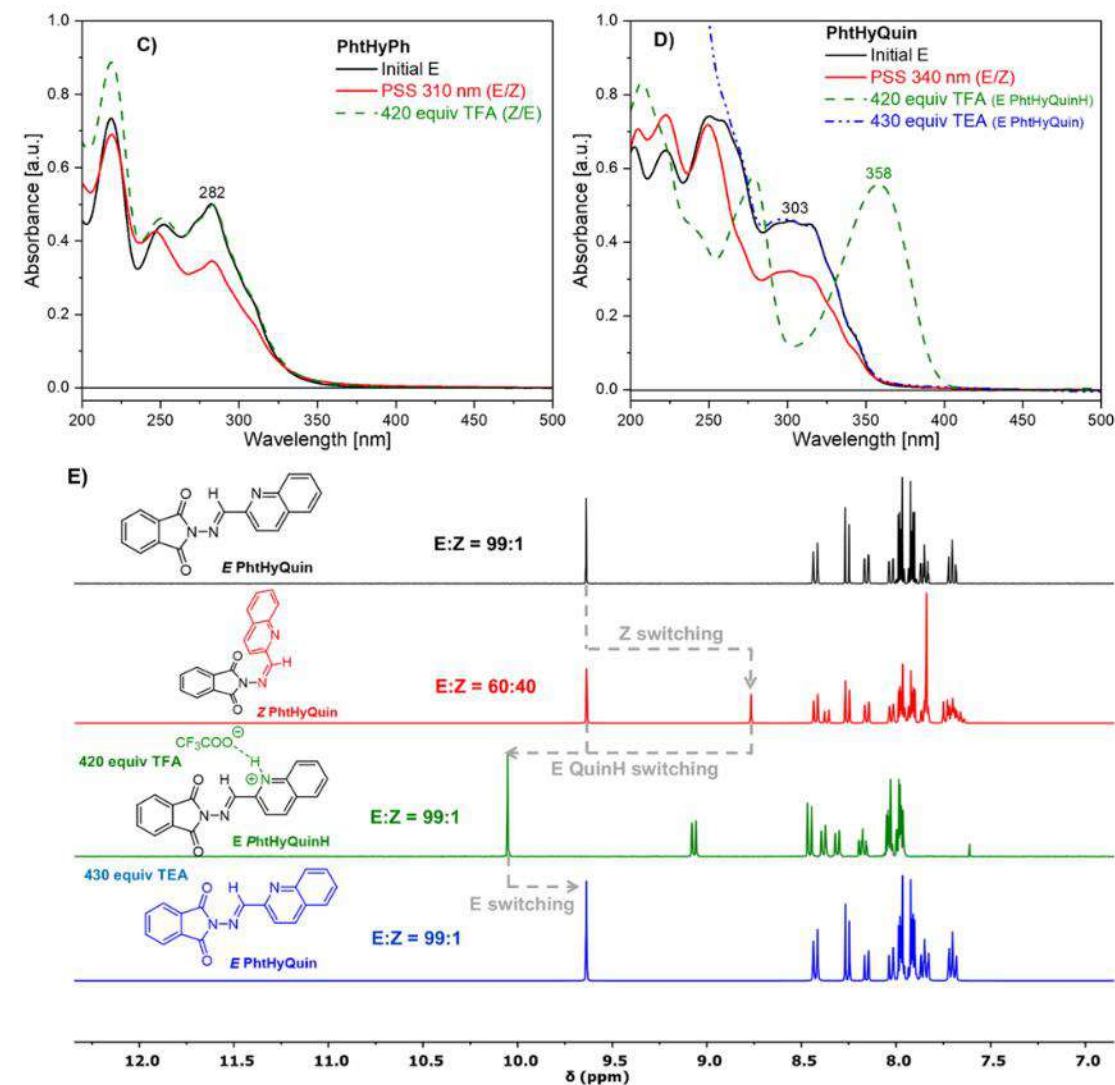
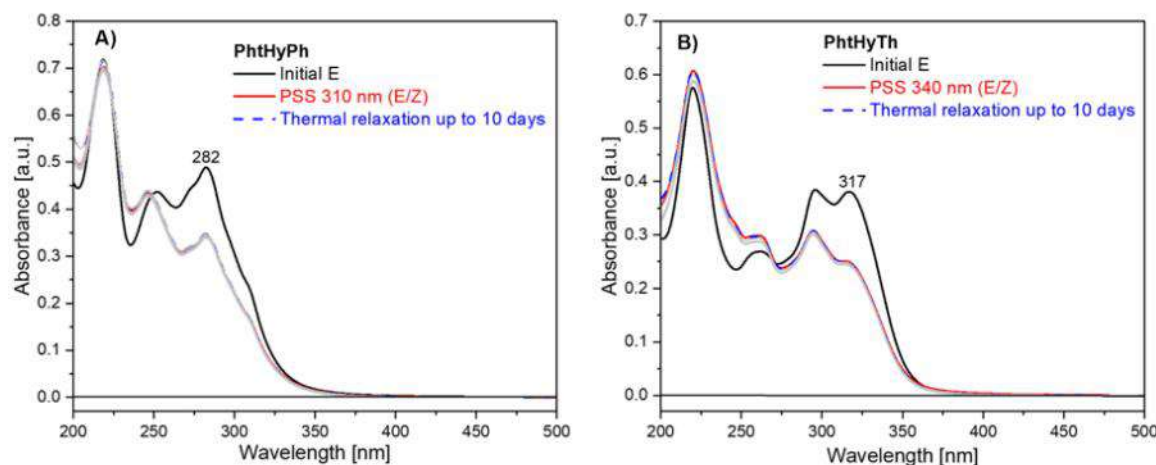
Determinants of Z-Isomer Stability in Phthalylhydrazones: Dual Light and Acid Control

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The results are developed as part of contract №: BG-RRP-2.004-0002-C01, **Laboratory of Organic Functional Materials** (Project BiOrgaMCT), Procedure BG-RRP-2.004 “Establishing of a network of research higher education institutions in Bulgaria”, funded by BULGARIAN NATIONAL RECOVERY AND RESILIENCE PLAN